

Efficient Recognition of an Unpaired Lesion by a DNA Repair Glycosylase

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The double-helical structure of DNA protects the nucleobases from oxidative and alkylative chemical damage.¹ However, this internal base pairing is also a barrier to the enzymes that recognize and repair DNA damage.² Enzymes that modify bases in DNA use nucleotide flipping to rotate the target nucleotide out of the duplex into an active site, but the energetic barrier that must be overcome is not fully understood. Unpaired (bulged) nucleotides are more accessible, but it is not known to what extent they are recognized by nucleotide-flipping enzymes. We have investigated this question with human alkyladenine DNA glycosylase (AAG). AAG recognizes a wide variety of structurally disparate lesions, including deoxyinosine (I), which results from the oxidative deamination of adenosine. AAG catalyzes the hydrolysis of the *N*-glycosidic bond to release the lesion base and initiate the base excision repair pathway (Figure 1A).³ We used single-turnover kinetics to demonstrate that AAG is exquisitely sensitive to the structural context of the I lesion. An inverse correlation between duplex stability and catalytic efficiency was observed, indicating that stable pairing is a barrier to recognition by AAG. Single-stranded DNA is a very poor substrate, but a single-nucleotide bulge is recognized more efficiently than any other context. These results highlight the intrinsic barrier to nucleotide flipping that DNA repair enzymes face and imply that the recognition of DNA damage by AAG is remarkably plastic.

Crystal structures of AAG bound to an extrahelical lesion have revealed that the DNA is bent, the lesioned base is rotated by $\sim 180^\circ$ out of the duplex into the active-site pocket, and the hole that is left in the DNA is plugged with the phenolic group of Y162 (Figure 1B).⁴ Intimate protein–DNA contacts in this static structure seem to preclude the possibility that a bulge could be accommodated. However, the barrier to flipping of a bulged nucleotide is much less than that for a mismatch. Bulged purines favor an intercalated conformation, but an extrahelical conformation has been observed in a crystal structure.⁵ If AAG has sufficient flexibility to encompass a bulged nucleotide, then the decreased barrier to flipping would enable more efficient recognition.

Therefore, we characterized the activity of AAG toward an I lesion in different structural contexts, including a single-nucleotide bulge. Product release is rate-limiting for multiple turnover reactions of AAG, requiring the use of single-turnover methods.⁶ The concentration of AAG was varied to determine the maximal single-turnover rate constant (k_{\max}) and the half-maximal concentration ($K_{1/2}$) for each context (Figure 2). Although the individual reaction rates varied by more than 10 000-fold, with half-times that varied from seconds to days, single-exponential fits were excellent in all cases ($R^2 > 0.97$; see the Supporting Information for representative data). The rate constant k_{\max} reports on the flipping and *N*-glycosidic bond hydrolysis steps, and the $K_{1/2}$ value corresponds to the K_d for substrate dissociation.⁶ The specificity between competing substrates is given by the ratio $k_{\max}/K_{1/2}$, the catalytic efficiency (analogous to k_{cat}/K_M in steady-state kinetics).

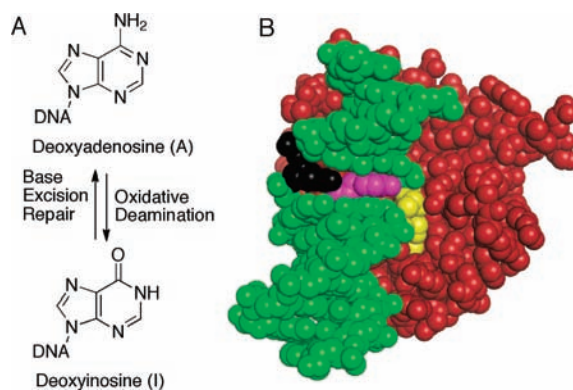


Figure 1. Deamination of A to form I and the structure of AAG bound to damaged DNA. (A) Oxidative deamination of A is reversed by AAG and the base excision repair pathway. (B) AAG is red, the intercalating tyrosine (Y162) is magenta, and the DNA is green, except for the lesion (yellow) that is flipped into the active site and the opposing nucleotide (black). This opposing nucleotide is missing in the bulged substrate. Coordinates are from the Protein Data Bank structure for AAG bound to 1,*N*⁶-ethenoA-DNA (PDB entry 1EWN).⁴

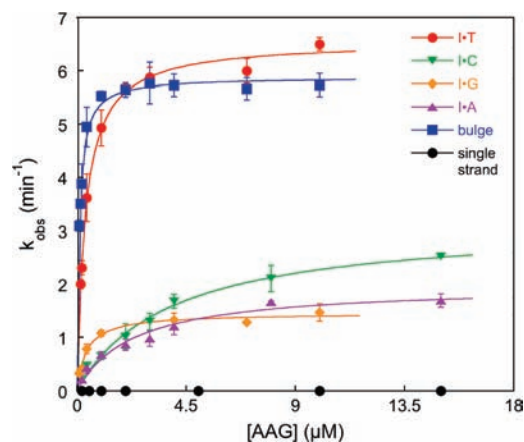


Figure 2. Concentration dependence for single-turnover glycosylase activity of AAG. Each data point corresponds to the average and standard deviation from 4–8 individual reactions (see the Supporting Information). Lines indicate the best fits to a hyperbolic concentration dependence: $k_{\text{obs}} = k_{\max}[\text{AAG}]/(K_{1/2} + [\text{AAG}])$.

The glycosylase activity of AAG is shown in Figure 2 and summarized in Table 1. The bulge was the best context tested, with a catalytic efficiency that is 3-fold better than that for the biological I·T mismatch. This result indicates that the increased ease in flipping the lesion more than compensates for any unfavorable effects of removing the opposing base (Figure 1B). Further work is needed to evaluate whether the bulge might be more easily bent than the mismatch and whether this could contribute to catalytic recognition. Other mismatches were recognized less efficiently than the I·T mismatch. The single-stranded lesion is even more

Table 1. Kinetic Parameters for the AAG-Catalyzed Hydrolysis of Inosine in Different Structural Contexts^a

opposing base	$K_{1/2}$ (μM)	k_{max} (min^{-1})	$k_{\text{max}}/K_{1/2}$ ($\text{M}^{-1} \text{s}^{-1}$) ^b	relative $k_{\text{max}}/K_{1/2}$
bulge (none)	0.08	5.4	1.1×10^6	3
T	0.30	6.3	3.5×10^5	(1)
G	0.41	1.5	6.2×10^4	0.2
C	3.8	3.3	1.4×10^4	0.04
A	2.3	2.0	1.4×10^4	0.04
single strand ^c	0.34	0.0035	1.7×10^2	0.0005

^a Data were obtained at 23 °C in 50 mM NaMES (pH 6.1), 1 mM EDTA, 1 mM DTT, 0.1 mg/mL BSA. The ionic strength was adjusted to 200 mM with NaCl. ^b $k_{\text{max}}/K_{1/2}$ is a measure of the catalytic efficiency, analogous to $k_{\text{cat}}/K_{\text{M}}$ in steady-state kinetics. ^c This oligonucleotide does not appear to adopt secondary structure, because a single-stranded polyT oligonucleotide gave very similar kinetic parameters (See Supporting Information).

accessible, and several glycosylases efficiently utilize single-stranded substrates; however, we found that AAG has ~2000-fold reduced activity toward this substrate.

Since AAG does not make specific contacts with the opposing base, the relationship between base-pair stability and catalytic efficiency for different mismatches can be assessed. Thermodynamic parameters for duplex stability for inosine paired with each of the normal nucleotides are known⁷ and serve as a surrogate for base-pair stability. The relative free energies ($\Delta\Delta G$) for catalytic efficiency of the AAG-catalyzed reaction and for duplex stability are presented in Figure 3. A linear fit to all of the mismatches yields a slope of -0.96 ($R^2 = 0.67$), and omission of the I•T substrate gives a slope of -0.59 ($R^2 = 0.97$). This is a limited set of data, but the trend is clear. Previous analysis of the thermodynamics of duplex stability and DNA binding by uracil DNA glycosylase found that disruption of base pairing gives greater accessibility and tighter binding.⁸ The inverse relationship between duplex stability and efficiency of excision supports the model that AAG and other glycosylases must overcome the barrier provided by base-pairing interactions. This behavior maximizes the discrimination between damaged and undamaged nucleotides because undamaged nucleotides have more favorable hydrogen-bonding and stacking interactions.

The I•T mismatch deviates by ~1 kcal/mol relative to the other mismatches, raising the possibility that AAG recognizes the I•T wobble-pair geometry independent of its effects on duplex stability. Indeed, I•G forms a less stable duplex than I•T and yet is removed with 5-fold lower catalytic efficiency. Deamination of A in DNA generates an I•T pair, and AAG-initiated repair restores the correct sequence. Incorporation of dIMP or replication of an I lesion predominantly forms an I•C pair.⁹ Under these scenarios, the activity of AAG would make a permanent mutation. The much lower efficiency of AAG toward I•C leaves open the possibility that another DNA repair pathway corrects replicative events.

By quantifying the energetic differences in the catalytic efficiency of AAG with different mismatches, we have obtained strong evidence that base pairing provides a barrier to base excision. The fact that a bulge is recognized with the same efficiency as a mismatch indicates considerable flexibility in DNA recognition. This also has biological ramifications, as DNA polymerases can slip on repetitive or damaged templates to generate bulged

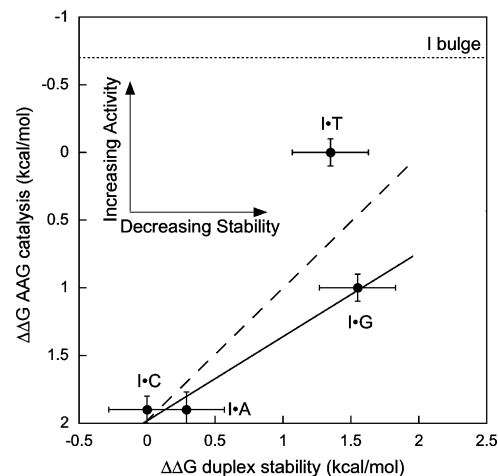


Figure 3. Linear free-energy relationship showing an inverse correlation between duplex stability and glycosylase activity. Differences in free energy ($\Delta\Delta G$) are from ref 7 for duplex stability and from the equation $\Delta\Delta G = -RT \ln(k_{\text{max}}/K_{1/2}^{\text{rel}})$ for AAG activity. Linear fits to all of the mismatches (dashed line; slope = -0.97 , $R^2 = 0.67$) and with the exclusion of the T mismatch (solid line; slope = -0.59 , $R^2 = 0.97$) are shown. The relative activity toward a bulge is shown as a dotted line.

structures. Initiation of base excision repair could correct nascent +1 frameshifts but would make -1 frameshifts permanent. This ability of AAG to act on bulged nucleotides may explain the observation that an increased level of AAG expression is correlated with increased frequency of frameshift mutations.¹⁰

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Supporting Information Available: Materials and methods, representative kinetic data, glycosylase activity with a different single-stranded oligo, and supporting results and discussion. This material is available free of charge via the Internet at <http://pubs.acs.org>.

References

- (1) Lindahl, T. *Nature* **1993**, *362*, 709.
- (2) Roberts, R. J.; Cheng, X. *Annu. Rev. Biochem.* **1998**, *67*, 181.
- (3) Stivers, J. T.; Jiang, Y. L. *Chem. Rev.* **2003**, *103*, 2729.
- (4) Lau, A. Y.; Wyatt, M. D.; Glassner, B. J.; Samson, L. D.; Ellenberger, T. *Proc. Natl. Acad. Sci. U.S.A.* **2000**, *97*, 13573.
- (5) (a) Patel, D. J.; Kozłowski, S. A.; Marky, L. A.; Rice, J. A.; Broka, C.; Itakura, K.; Breslauer, K. J. *Biochemistry* **1982**, *21*, 445. (b) Joshua-Tor, L.; Rabinovich, D.; Hope, H.; Frolow, F.; Appella, E.; Sussman, J. L. *Nature* **1988**, *334*, 82.
- (6) (a) Abner, C. W.; Lau, A. Y.; Ellenberger, T.; Bloom, L. B. *J. Biol. Chem.* **2001**, *276*, 13379. (b) Baldwin, M. R.; O'Brien, P. J. *Biochemistry* **2009**, *48*, 6022.
- (7) Watkins, N. E., Jr.; SantaLucia, J., Jr. *Nucleic Acids Res.* **2005**, *33*, 6258.
- (8) (a) Krosky, D. J.; Schwarz, F. P.; Stivers, J. T. *Biochemistry* **2004**, *43*, 4188. (b) Krosky, D. J.; Song, F.; Stivers, J. T. *Biochemistry* **2005**, *44*, 5949.
- (9) (a) Yasui, M.; Suenaga, E.; Koyama, N.; Masutani, C.; Hanaoka, F.; Gruz, P.; Shibutani, S.; Nohmi, T.; Hayashi, M.; Honma, M. *J. Mol. Biol.* **2008**, *377*, 1015. (b) Zhang, H.; Bren, U.; Kozekov, I. D.; Rizzo, C. J.; Stec, D. F.; Guengerich, F. P. *J. Mol. Biol.* **2009**, *392*, 251.
- (10) Hofseth, L. J.; Khan, M. A.; Ambrose, M.; Nikolayeva, O.; Xu-Welliver, M.; Kartalou, M.; Hussain, S. P.; Roth, R. B.; Zhou, X.; Mechanic, L. E.; Zurer, I.; Rotter, V.; Samson, L. D.; Harris, C. C. *J. Clin. Invest.* **2003**, *112*, 1887.

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